## Anionic Fries Rearrangements of Esters of *ortho*-lodobenzyl Alcohols: Rapid Routes to Oestrone Methyl Ether and Its $9\beta$ Epimer, and Aryl Naphthalide Lignans

## Stephen Horne and Russell Rodrigo\*

Guelph–Waterloo Center for Graduate Work in Chemistry, Department of Chemistry, University of Waterloo, Waterloo, Ontario N2L 3G1, Canada

A fast, general, low-temperature rearrangement of *ortho*-iodobenzyl esters, triggered by lithium-iodine exchange, leads to isobenzofurans which are intercepted *in situ* by inter and intramolecular Diels-Alder (IMDA) reactions to produce a variety of carbocycles including natural lignans and steroids.

The metal-halogen exchange-initiated intramolecular cycliacylation and cyclialkylation of aromatic halides bearing appropriate electrophilic centres is a useful method for the preparation of carbocycles and heterocycles. 1 Documented examples of successful exchange-initiated cyclizations of bromo- or iodo-carbonyl substrates, which are potentially enolizable by the initially generated nucleophilic centre, however, are relatively scarce.<sup>2</sup> Our recent investigations into anionic variants of the Fries rearrangement as applied to regiospecific xanthone<sup>3</sup> and acridone<sup>4</sup> synthesis, coupled with a recent disclosure<sup>5</sup> concerning the exchange-initiated conversion of the non-enolizable bromopivaloate 1 to the hydroxyphthalan 2 ( $R = Bu^t$ ), prompted us to enquire as to whether such a homologous Fries rearrangement might be generally applicable to substrates such as 3 in which the acid derived R substituent possessed potentially acidic and/or reactive (but subsequently useful) functional groups. Success in such a manoeuvre would require a rate of lithium-iodine exchage in 3 not only rapid enough to supersede the rate of competitive deprotonation by the alkyllithium reagent but also a rate of subsequent 5-exo-trig cyclization to hydroxyphthalans 2 which would preempt the intervention of any kinetically favoured deprotonation alternatives. We now report the development of this as a general approach to substituted hydroxyphthalans 2, known precursors to isobenzofurans, useful in brief preparations of many natural products.6 Condensation of 2-iodobenzyl alcohol with 4-pentenoic acid 1,3-dicyclohexylcarbodiimide [(DCC), 4-dimethylaminopyridine (DMAP), CH<sub>2</sub>Cl<sub>2</sub>] afforded benzyl ester **3a** (91%). Treatment of a 0.07 mol dm<sup>-3</sup> solution of 3a in THF-Et<sub>2</sub>O-hexanes (4:1:1) with one equivalent of LiBu at -100 °C followed by immediate quenching with ammonium chloride and work-up afforded

Br 
$$R^2$$
  $O$   $R^1$   $R^2$   $O$   $R^1$   $R^2$   $O$   $R^3$   $R^4$   $R$ 

**3a**;  $R^1 = (CH_2)_2CH = CH_2$ ,  $R^2 = R^3 = R^4 = H$ **b**;  $R^1 = 3,4$ -(methylenedioxy)phenyl,  $R^2 = H$ ,  $R^3 = R^4 = OMe$ 

c;  $R^1 = (CH_2)_3C \equiv CH$ ,  $R^2 = R^3 = R^4 = H$ 

**d**;  $R^1 = (CH_2)_3C \equiv CTMS$ ,  $R^2 = R^3 = R^4 = H$ 

e;  $R^1 = (CH_2)_4CH=CH_2$ ,  $R^2 = R^3 = R^4 = H$ 

 $f; R^1 = Ph, R^2 (CH_2)_3 CH = CH_2, R^3 = R^4 = OMe$ 

 $g; R^1 = Et, R^2 = (CH_2)_3CH=CH_2, R^3 = R^4 = OMe$ 

 $\mathbf{h}$ ; R<sup>1</sup> = Me, R<sup>2</sup> = (CH<sub>2</sub>)<sub>3</sub>CH=CH<sub>2</sub>, R<sup>3</sup> = R<sup>4</sup> = OMe

i;  $R^1 = CH_2OCH_2CH=CH_2$ ,  $R^2 = R^3 = R^4 = H$ 

$$R^2$$
 $R^3$ 
 $CO_2Me$ 
 $R^1$ 
 $CO_2Me$ 
 $R^1$ 
 $CO_2Me$ 
 $R^2$ 
 $CO_2Me$ 
 $R^3$ 
 $CO_2Me$ 
 $R^3$ 
 $R^4 = (CH_2)_2CH=CH_2, R^2 = R^3 = H$ 
 $R^4$ 
 $R^4$ 

crude material devoid of any aromatic iodine. Treatment of this crude product with dimethyl acetylene dicarboxylate (cat. HOAc, 100 °C, 30 min) furnished intermolecular Diels–Alder adduct **4a**<sup>7</sup> in 60% yield. Similar treatment of ester **3b**<sup>8</sup> resulted in the formation of **4b** (42% overall from **3b**), which had previously been prepared<sup>9</sup> and regiospecifically converted into the arylnaphthalide lignans diphyllin **5a** and justicidin A **5b**. <sup>9,10</sup>

Of particular interest in the current study was the possibility of incorporating into esters 3 a tethered dienophilic residue for subsequent use in intramolecular Diels-Alder processes. Individual exposure of esters 3c and d to the usual rearrangement conditions followed by refluxing of the crude rearrangement products in benzene (cat. HOAc, 16 h) afforded adducts 6a (34%) and 6b (38%) in comparable yield. Thus, protonation of the aryllithium by the free acetylenic hydrogen does not appear to be a significant process. Similarly, ethylenic esters 3e-g afforded the  $exo^{11}$  adduct 77 (80%), 8a (64%) and 8b (72%), respectively, of which only the latter readily aromatized to the naphthalene 9 on leaving to stand at 25 °C. Presumably the higher overall yields observed for these ethylenic-derived adducts over the acetylenic counterparts 6a and b is a consequence of more favourable orbital overlap between the transient IBFs and the ethylenic dienophiles, an idea supported by molecular models. The flexibility of the overall process is obvious in that the dienophilic residue for the IMDA reaction can be tethered to the acyl portion (as in 3c and d) or the alcohol portion (as in 3f and g) of the starting iodobenzyl ester. The latter starting materials were prepared by addition of the Grignard reagent derived from 5-bromopentene to 2-iodo-4,5-dimethoxybenzaldehyde8 followed by esterification of the resulting secondary alcohol. It is interesting that the rearrangement/Diels-Alder process of acetate 3h and allyloxyacetate 3i did not lead to either the isolation of 8c or 10, respectively. The higher kinetic acidity of the methyl

group **3h** and the oxymethylene group **3i** are probably responsible for the outcome—the recovery of the deiodinated starting materials.

16

The synthetic utility of the rearrangement–IMDA sequence is demonstrated here by a rapid total synthesis of  $(\pm)$ -oestrone methyl ether, a synthetic target of several previous IMDA approaches employing o-quinodimethanes. 12 Pauson–Khand reaction<sup>13</sup> of methyl pent-4-ynoate with Co<sub>2</sub>(CO)<sub>8</sub> and ethylene [65 °C, 150 psi (psi = 6.9 kPa) 6 h, toluene] afforded 11 (29%). Conjugate addition of vinylmagnesium bromide (cat. CuI, Me<sub>2</sub>S, THF, -60 °C) followed by enolate trapping (MeI, HMPA, -40 °C) afforded 12a and b, (71%, 2:1 diastereoselectivity) in which the major diastereoisomer 12a possessed the two larger vicinal groups in a trans relationship.14 Ketalization of the mixture 12a, b (ethylene glycol, TsOH, benzene, reflux) afforded the separable diastereoisomers 13a and b. Transesterification of separated 13a and b (NaOH-MeOH, then DCC, DMAP, 2-iodo-3-methoxybenzyl alcohol)† provided 14a and b. Exposure of 14a to the standard rearrangement-IMDA sequence afforded a single crystalline

<sup>†</sup> Prepared in 42% yield by iodination (Ag<sub>2</sub>SO<sub>4</sub>, CH<sub>2</sub>Cl<sub>2</sub>, I<sub>2</sub>, 25 °C, 5 h) of 3-methoxy benzyl alcohol.

adduct 15‡ in 27% yield by the favoured transition state 15a involving a pseudo-chair-like C ring and exo-β-face delivery<sup>12a</sup> of the dienophile. The production of significant amounts of 3-methoxybenzyl alcohol (presumably formed via ester cleavage) during the rearrangement step accounted for the bulk of the material balance. This tendency for ester cleavage had not been observed in any of our prior experiments but attempted conversion of 14b to the corresponding C-14 epimer of 15 failed completely for the same reason. Hydrogenolysis of 15 (AcOH, 50 psi, 3 h, 10% Pd/C) followed directly by deketalization (AcOH, THF, H2O, heat) afforded in 60% total yield a 1:2 mixture of ( $\pm$ )-oestrone methyl ether 16 and its C-9 epimer as estimated by 500 MHz <sup>1</sup>H NMR spectroscopy in comparison with an authentic sample of 16. We have not been able to separate the C-9 epimers, even by reverse-phase chromatography. The epimeric mixture results from partial inversion of configuration at C-9 during reductive removal of the oxygen bridge in 15. Although this was not unexpected,15 hydrogenolysis of a similar bridged intermediate had been reported<sup>11</sup> to proceed with complete retention of configuration at C-9. More examples of the rearrangement-IMDA sequence, especially with nitrogen-containing dienophiles, are being sought and modifications designed to forestall the ester cleavage and improve stereocontrol at C-9

‡ Selected spectroscopic data for 15: m.p. 145–146 °C (Et<sub>2</sub>O–hexane); ¹H NMR (250 MHz, CDCl<sub>3</sub>):  $\delta$  0.95 (s, 3H, Me), 1.19–2.52 (m, 12H, 2 H-16, 2 H-15, 2 H-7, 2 H-11, 2 H-12, H-8, H-14), 3.78 (s, 3H, OMe), 3.81–3.96 (m, 4H, ketal CH<sub>2</sub>CH<sub>2</sub>), 5.24 (d, J 5.0, Hz 1H, H-6), 6.65 (dd, J 8.0 Hz, 2.2, 1H, H-2), 6.78 (d, J 2.2 Hz, 1H, H-4), 7.02 (d, J 8.0 Hz, 1H, H-1); MS (EI) mlz 342 (14, M<sup>+</sup>), 174 (81), 123 (12), 99 (100), 86 (15), 55 (4), 40 (28); HRMS (EI) Calc. for C<sub>21</sub>H<sub>26</sub>O<sub>4</sub>: 342.1832. Found: 342.1826.

during hydrogenolysis are contemplated. Results of these investigations will be reported elsewhere.

We thank the NSERC of Canada for support of this work.

Received, 10th October 1991; Com. 1/05164D

## References

- 1 W. E. Parham and C. K. Bradsher, Acc. Chem. Res., 1982, 15, 300.
- 2 M. P. Cooke, Jr. and R. K. Widener, J. Org. Chem., 1987, 52, 1381 and references cited therein.
- 3 S. Horne and R. Rodrigo, J. Org. Chem., 1990, 55, 4520.
- 4 S. Horne and R. Rodrigo, J. Chem. Soc., Chem. Commun., 1991, 1046.
- 5 J. A. Miller, J. Org. Chem., 1987, 52, 323.
- 6 R. Rodrigo, Tetrahedron, 1988, 44, 2093.
- 7 S. K. Meegalla and R. Rodrigo, Synthesis, 1989, 942.
- 8 Prepared by NaBH<sub>4</sub> reduction of 2-iodo-4,5-dimethoxybenz-aldehyde and subsequent esterification with piperonylic acid, D. M. Piatak, G. Flynn, K. Yim and J. Roosenberg, *J. Org. Chem.*, 1977, **42**, 1069.
- B. A. Keay, H. P. Plaumann, D. Rajapaksa and R. Rodrigo, Can. J. Chem., 1983, 61, 1987.
- 10 H. P. Plaumann, J. G. Smith and R. Rodrigo, J. Chem. Soc., Chem. Commun., 1980, 354.
- 11 W. Friedrichsen, B. M. Konig, K. Hildebrandt and T. Debaerdemaeker, *Heterocycles*, 1986, 24, 297.
- 12 (a) R. L. Funk and K. P. C. Vollhardt, Chem. Soc. Rev., 1980, 9, 41; (b) T. Kametani and H. Nemoto, Tetrahedron, 1980, 37, 3; (c)
  W. Oppolzer, Synthesis, 1978, 793; (d) D. Taub, in The Total Synthesis of Natural Products, ed. J. ApSimon, Wiley, New York, 1984, vol. 6, pp. 16–37.
- 13 N. E. Schore, Chem. Rev., 1988, 88, 1081.
- 14 T. Takahashi, M. Nisar, K. Shimizu and T. Tsuji, *Tetrahedron Lett.*, 1986, 27, 5103.
- 15 P. N. Rylander, Catalytic Hydrogenation in Organic Syntheses, Acaedemic Press, New York, 1979, pp. 271–274.